

## Naming Covalent Compounds Worksheet

Write the formulas for the following covalent compounds:

- 1) nitrogen tribromide  $NBr_3$
- 2) hexaboron monosilicide  $B_6Si$
- 3) chlorine dioxide  $ClO_2$
- 4) hydrogen monoiodide  $HI$
- 5) iodine pentafluoride  $IF_5$
- 6) dinitrogen trioxide  $N_2O_3$
- 7) ammonia (nitrogen trihydride)  $NH_3$
- 8) phosphorus triiodide  $PI_3$
- 9) dihydrogen monoxide  $H_2O$
- 10) diphosphorous pentoxide  $P_2O_5$

Write the names for the following covalent compounds:

- 11)  $P_4S_5$  tetra phosphorus pentasulfide
- 12)  $O_2$  oxygen gas
- 13)  $SF_6$  sulfur hexafluoride
- 14)  $Si_2Br_6$  disilicon hexabromide
- 15)  $SCl_4$  sulfur tetrachloride
- 16)  $CH_4$  carbon tetrahydride
- 17)  $B_2Si$  digerman silicide
- 18)  $NF_3$  nitrogen trifluoride
- 19)  $H_2O$  water
- 20)  $N_2O_5$  dinitrogen pentoxide